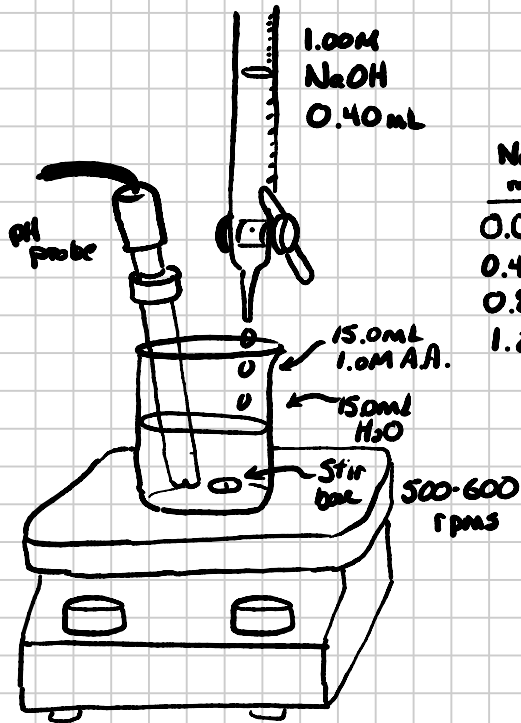
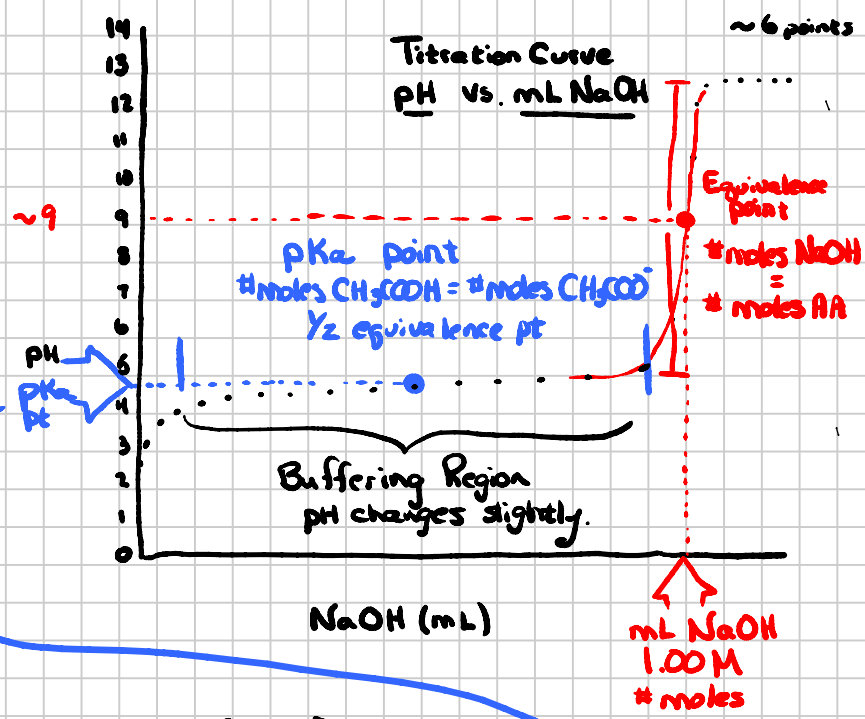


Titration



NaOH ml	pH
0.00	~
0.40	~
0.80	~
1.20	~



$$pH = -\log [H_3O^+]$$

$$pOH = -\log [OH^-]$$

$$pKa = -\log K_a \dots -K_a = 10^{-pKa}$$