

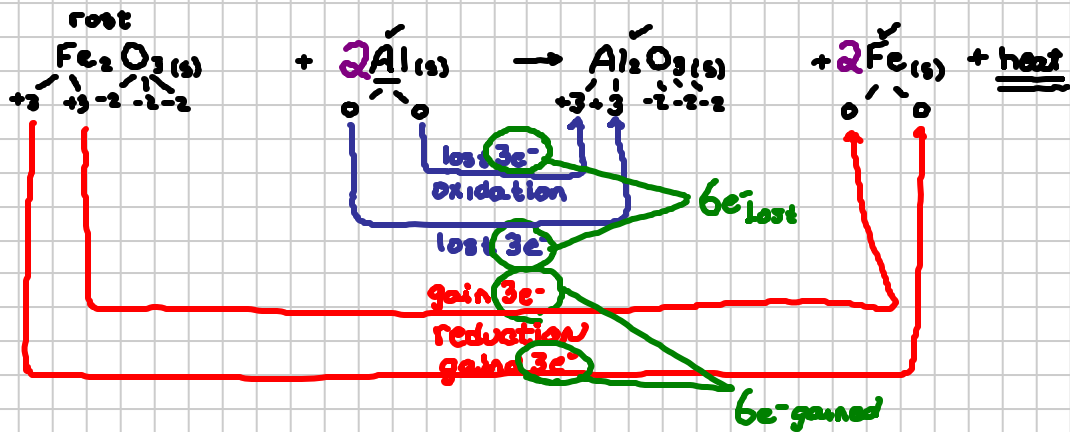
Lecture 20 2 Redox Reactions

Note Title

11/18/2011

Oxidation-Reduction Reactions: Δ O.N, e^- movement

Thermite Reaction



Redox Rxn: $e^-_{\text{lost}} = e^-_{\text{gained}}$

"Al is oxidized", "Fe is reduced"

Fe_2O_3 oxidizing agent, Al is the reducing agent