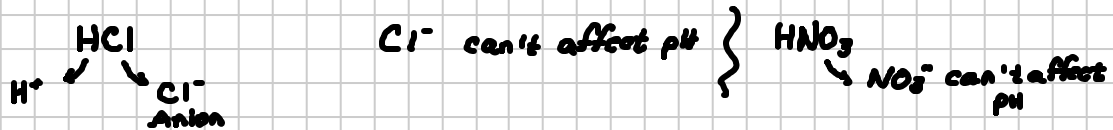


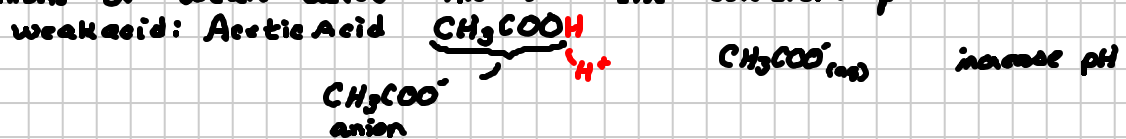
1) Anions of strong acids don't affect the solution's pH.



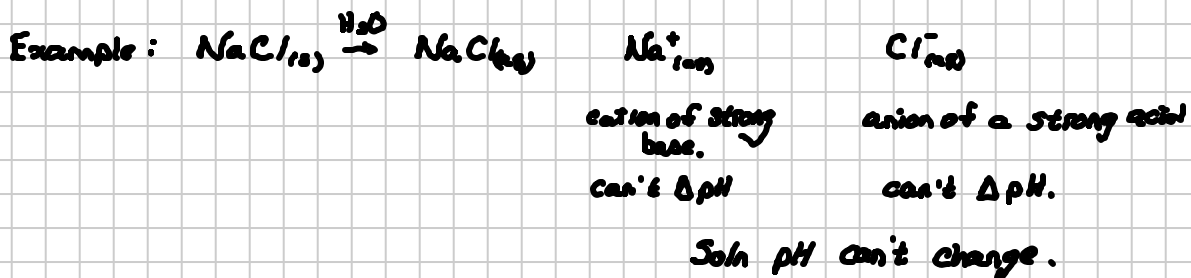
2) Cations of strong bases don't affect the solution's pH.



3) Anions of weak acids increase the solution's pH



4) Cations that are conjugate acids of weak bases lower the soln's pH.



Example: Sodium Acetate NaCH_3COO

