Electronegativity: Electron Distribution in a Bond



Electron density shifts in the direction of the more electronegative atom (oxygen).



Electronegativity: Electron Distribution in a Bond

Figure 9.19 in text.MethanolE.N. = 3.5E.N. = 2.1E.N. = 3.5 > E.N. = 2.1More E.N.Less E.N.Less E.N.The O-H bond is a polar bond because it has + and – ends!

Dipole Moment Vector:

...an arrow that indicates the direction electrons shift.

...includes an additional mark to indicate positive end.

...usually drawn through the bond.



